

## **Droplet Stability and Coalescence**

A problem for the Twenty-Eighth Annual Workshop on Mathematical Problems in Industry

At the University of Delaware

Mark Hurwitz June 11, 2012

**Better Lives. Better Planet.sm** 

## About Pall Corporation

- Leader in Filtration, Separation and Purification
- Scientific, technology and advanced engineering company
- Founded in 1946
- Global Footprint
- Over 10,000 Employees
- \$2.7B Annual Revenue year ending Aug. 2011



**Droplet Stability** 

Mark Hurwitz

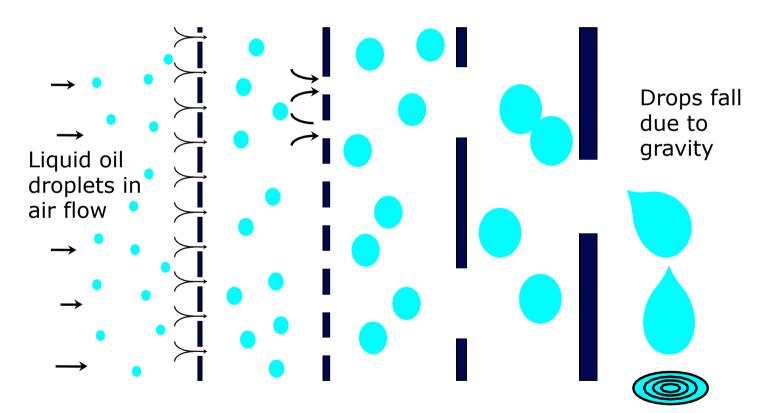






## Coalescence: Two or more droplets merge

- Random collisions can cause coalescence
- A coalescer forces droplet interaction in a fibrous structure
- Schematic example:





## Big equipment means small improvements are important



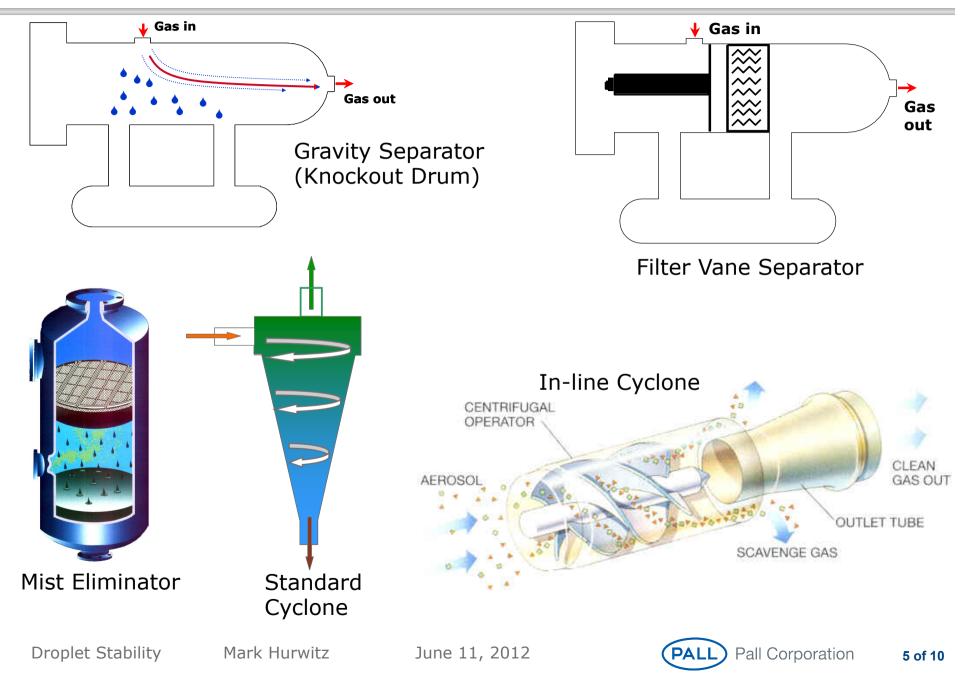
North Sea off-shore gas field. Removal of salt-containing produced water before export compressor



Middle East gas treating plant. Removal of amine from sweet gas downstream of contactor, protection of glycol unit



## Other types of liquid / gas separators

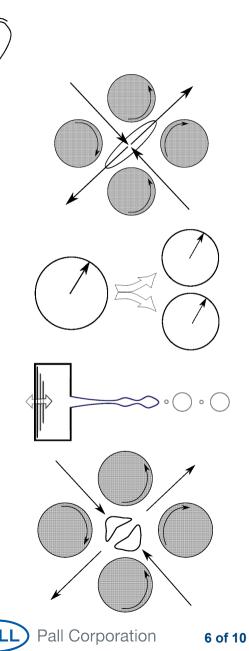


## Brief ( and very select) history of droplet analysis

- **1932,** H. Lamb, "Hydrodynamics, 6<sup>th</sup> edition"
  - Inertia versus gravity for waves and tides
- 1934, G. I. Taylor, "The Formation of Emulsions in Definable Fields of Flow"
  - Viscosity versus surface energy
  - Emulsions and energy to make small droplets
- **1960,** R. Shinnar & J. M. Church, "Predicting Particle Size in Agitated Dispersions"
  - Surface energy versus turbulence
  - Estimate stable droplet size in turbulent flow
- **1994,** J. Eggers & T. F. Dupont, "Drop formation in a one-dimensional approximation of the Navier-Stokes equation"
  - Surface energy, inertia and viscosity
  - Ink jet printing jet stability and drop size
- 2006, F. Baldessari & L. G. Leal, "Effect of overall drop deformation on flow-induced coalescence at low capillary numbers
  - Coalescence of viscous droplets in shear flow

Droplet Stability

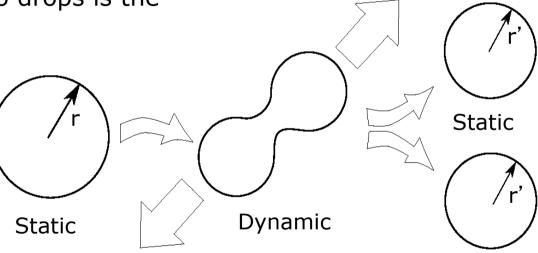
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## Problem 1: What makes an inviscid drop be stable?

- Reasonable starting point:
- Radius: r = 0.1 to 10 cm
- Constant density:  $\rho = 1.0$  gm/c.c.
- Surface tension:  $\sigma = 70$  dyne/cm
- Neglect viscosity and evaporation
- Minimum energy to make two drops is the change in surface energy
- Is this a useful criterion?

• Volume conservation: - r' = r /  $2^{1/3}$ 



- Change in surface energy:
  - $-\Delta E = 4\pi\sigma (2r'^2 r^2) = 1.04\pi\sigma r^2$
  - Area increase = 25%

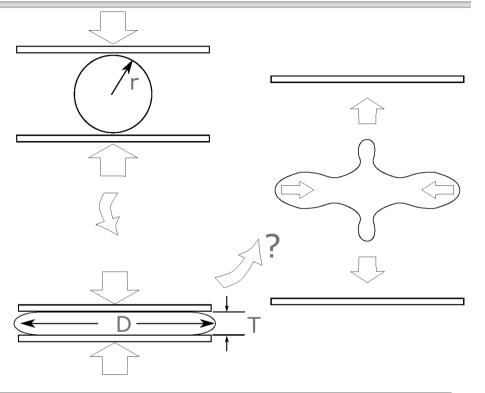
June 11, 2012



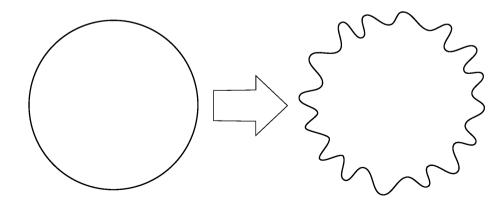
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## Surface area extremes

- Flatten droplet with parallel, non wetting plates
  - 25% increased area: T/2r ~0.49
  - Is this enough to expect instability?



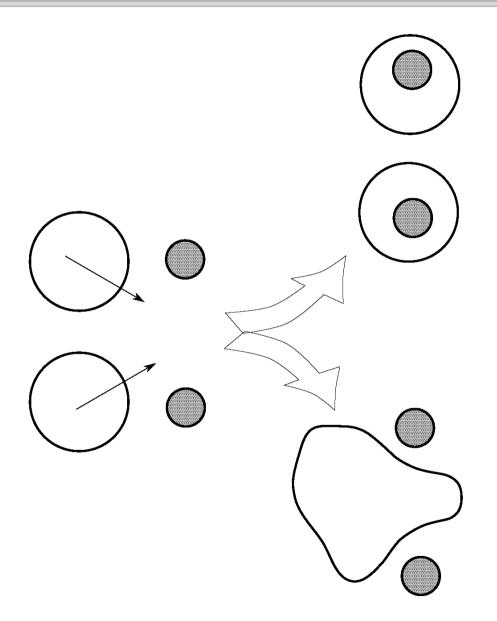
- Short wavelength pressure fluctuations
  - Large area increase with small change in aspect ratio
  - What are the resonances?





## Problem 2: Lubrication layer, phobic and phylic fibers

- (In case problem 1 is too easy!)
- Lubrication layer tends to keep droplets separate and separate from fibers
- Droplets wet and stick to phylic fibers
- Phobic fibers push droplets together
- What are the criteria that predict coalescence?





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# Thank you. And now the fun begins!

